

Columnar mesophases in supramolecular aggregates melamine-metallocacids: Liquid crystals of Fe, Cr, Mo and W at room temperature

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There are many reports on the effects of intermolecular hydrogen bonding in the field liquid crystals.¹ However, the studies on supramolecular metal-complexes based on hydrogen bond, are scarce.²

In this communication, we present examples of supramolecular aggregates melamine-metallocacids of Fe, Cr, Mo and W (Figure 1). They have been synthesized by combining equimolar amounts of 2,4,6-triarylarnino-1,3,5-triazine and metalloc-acids [Fe(CO)₄(CNC₆H₄CO₂H)] or [M(CO)₅(CNC₆H₄CO₂H)] (M = Cr, Mo, W). All the supramolecular aggregates obtained display columnar mesophases at room temperature, promoted by columnar packing of 1:1 melamine-metallocacid supramolecules. The stability of the columnar hexagonal mesophases formed is high and the clearing temperatures decrease very regularly in the order Fe > Cr > Mo > W. The compounds show high thermal stability, even in the isotropic state.

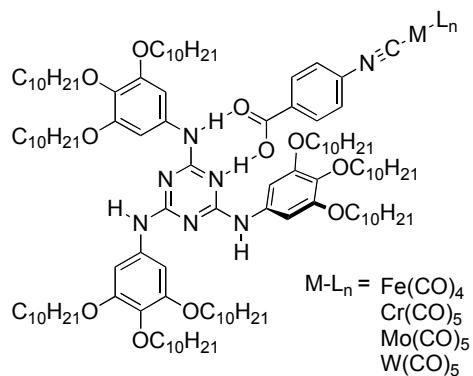


Figure 1

References

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